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# Principles Of Chemistry: A Molecular Approach Plus MasteringChemistry With EText -- Access Card Package (2nd Edition)





### Synopsis

ALERT: Â Before you purchase, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, Â you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products. Â Packages Access codes for Pearson's MyLab & Mastering products may not be included when purchasing or renting from companies other than Pearson; check with the seller before completing your purchase. Â Used or rental books If you rent or purchase a used book with an access code, the access code may have been redeemed previously and you may have to purchase a new access code. A Access codes Access codes that are purchased from sellers other than Pearson carry a higher risk of being either the wrong ISBN or a previously redeemed code. Check with the seller prior to purchase. Â -- Adapted from Nivaldo J. Troâ <sup>™</sup>s best-selling general chemistry text, Principles of Chemistry: A Molecular Approach focuses exclusively on the core concepts of general chemistry without sacrificing depth or relevance. Tro's unprecedented two- and three-column problem-solving approach is used throughout to give students sufficient practice in this fundamental skill. A unique integration of macroscopic, molecular, and symbolic illustrations help students to visualize the various dimensions of chemistry; and Troâ <sup>™</sup>s engaging writing style captures students&#39; attention with relevant applications. The Second Edition offers a wealth of new and revised problems, approximately 50 new conceptual connections, a new design and revised art program throughout, and is available with MasteringChemistry®, the most advanced online tutorial and assessment program available.

#### **Book Information**

Hardcover: 888 pages Publisher: Prentice Hall; 2 edition (December 26, 2011) Language: English ISBN-10: 0321750098 ISBN-13: 978-0321750099 Product Dimensions: 8.7 x 1.3 x 11.1 inches Shipping Weight: 5.5 pounds Average Customer Review: 4.2 out of 5 stars Â See all reviews (434 customer reviews) Best Sellers Rank: #801,867 in Books (See Top 100 in Books) #63 in Books > Science & Math > Chemistry > Molecular Chemistry #1013 in Books > Education & Teaching > Schools & Teaching > Instruction Methods > Science & Technology #2027 in Books > Science & Math > Chemistry > General & Reference

#### **Customer Reviews**

This is a new textbook for general chemistry. Yes, there a guite a number of general chemistry texts and they all cover roughly the same general material. They differ more in how they organize the material, illustrate it, and their method of helping the student develop a solid understanding and command of this basic material. The student not only needs to learn the basic facts of chemistry, she needs to learn how to think about it so when she looks at the world, she can see and understand the issues as they relate to this vital science.Dr. Tro begins this book by explaining how the text was developed and provides instructions for the student on how to make the most of the book. I liked this a great deal. He starts the book by providing what he thinks is the most important scientific idea in all of human knowledge: "the properties of matter are determined by the properties of molecules and atoms". From there he builds all the principles taught in general chemistry. He also define chemistry as: "the science that seeks to understand the behavior of matter by studying the behavior of atoms and molecules". Each chapter is organized in a similar fashion and a great deal of emphasis is placed on making the illustrations demonstrate a principle - a process - rather than an isolated picture. Tro also has interesting articles of chemistry in various fields in different chapters and he often introduces ideas with practical issues that will likely matter to the student. For review, at the end of each chapter he goes over the key concepts taught, the key equations and relationships discussed, and the key skills that the student is expected to have learned. He provides exercises to test one's mastery of the material that the student can answer for himself.

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